choice would have demanded mirror, twofold or inversion symmetry for the molecule. The polar space group  $Pca2_1$  was assumed and confirmed by the analysis. Refinement of the model and its enantiomorph yielded no significant differences in R, wR, goodness of fit or molecular dimensions. All H atoms were visible in difference maps and were included as riding atoms (C—H 0.95 Å) during the refinement.

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# Structure of 1,1,2,2,3,3-Hexamethyl-4-(2ethynylphenyl)-1,2,3-trigermacyclopent-4ene

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### Abstract

This is the first crystal structure determination of a member of the new germacyclic system of 1,2,3-trigermacyclopent-4-enes. The five-membered hetero-cycle is nearly planar with a maximum deviation of 0.139(10) Å from the least-squares plane through these

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Fig. 1. General view (*SHELXTL-Plus* graphic) of the molecule, showing the atom-numbering scheme. Anisotropic ellipsoids represent 50% probability boundaries. H atoms are represented as spheres of arbitrary radii.

#### Comment

The title compound is the first crystalline member of the fascinating germacyclic system of 1,2,3-trigermacyclopent-4-enes. The reaction and the recrystallization was carried out under dry argon because the crystals are air sensitive. The exact mechanism is unknown. A cyclotrigermane (Me<sub>2</sub>Ge)<sub>3</sub> is proposed as a reaction intermediate (Neumann, 1991). Ab initio quantum-mechanical determinations of the strain enthalpy of a hexamethylcyclotrigermane show a strong instability (Horner, Grev & Schäfer, 1992). This reactive intermediate reacts with 1,2-diethynylbenzene to give a trigermacyclopentene. The second ethynyl group does not react with another dimethylgermylene or cyclotrigermane. Presumably additional steric hindrance prevents access to the triple bond and prevents the reaction. Surprisingly the remarkable five-membered germaheterocycle is nearly planar. Evidence of this has also been provided by NMR spectroscopy (<sup>1</sup>H NMR: only three GeMe resonance signals).



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Lists of structure factors, anisotropic thermal parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 55594 (28 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: BR1013]

 $D_x = 1.567 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation  $\lambda = 0.71073 \text{ Å}$ 

reflections  $\theta = 6.4-14.7^{\circ}$   $\mu = 4.79 \text{ mm}^{-1}$  T = 170 KPlatelet

Colourless

 $R_{\text{int}} = 0.0314$   $\theta_{\text{max}} = 25.0^{\circ}$  $h = -9 \rightarrow 1$ 

 $k = -1 \rightarrow 18$ 

 $l = -21 \rightarrow 21$ 

6 standard reflections

frequency: 150 min intensity variation: none

Cell parameters from 16

 $0.44 \times 0.36 \times 0.14$  mm

#### Experimental

Crystal data
$C_{16}H_{24}Ge_3$
$M_r = 434.14$
Monoclinic
$P2_1/n$
a = 7.237 (7) Å
<i>b</i> = 14.952 (9) Å
<i>c</i> = 17.071 (20) Å
$\beta = 95.11 \ (8)^{\circ}$
$V = 1840 (3) \text{ Å}^3$
Z = 4

Data collection Nicolet R3m/V diffractometer  $\theta/2\theta$  scans Absorption correction: none 4379 measured reflections 3250 independent reflections 2297 observed reflections  $[F> 4.0 \sigma(F)]$ 

#### Refinement

 $(\Delta/\sigma)_{\rm max} = 0.001$ Refinement on FFinal R = 0.068 $\Delta \rho_{\rm max}$  = 2.1 e Å<sup>-3</sup> wR = 0.083 $\Delta \rho_{\rm min} = -2.6 \ {\rm e} \ {\rm \AA}^{-3}$ S = 0.87Atomic scattering factors 2297 reflections from International Tables 173 parameters for X-ray Crystallogra-Only H-atom U's refined phy (1974, Vol. IV, Table Calculated weights 2.2B)  $w=1/[\sigma^2(F)+0.007F^2]$ 

Data collection: Nicolet R3m/V software, release 4.11. Cell refinement: Nicolet R3m/V software. Data reduction: Nicolet R3m/V software. Program(s) used to solve structure: SHELXTL-Plus (Sheldrick, 1987). Program(s) used to refine structure: SHELXTL-Plus. Software used to prepare material for publication: PARST (Nardelli, 1983), PLATON (Spek, 1982), MISSYM (Le Page, 1987).

# Table 1. Fractional atomic coordinates and equivalent isotropic thermal parameters (Å<sup>2</sup>)

$U_{\rm eq} = \frac{1}{3} \sum_i \sum_j U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j.$					
	x	y	z	$U_{eq}$	
Ge(1)	0.07107 (14)	0.14797 (6)	0.43819 (5)	213	
Ge(2)	-0.04309 (14)	0.08593 (6)	0.31272 (6)	231	
Ge(3)	-0.02950 (14)	0.22783 (6)	0.24704 (5)	216	
C(1a)	-0.091 (2)	0.1300 (7)	0.5209 (6)	334	
C(1b)	0.322(1)	0.1147 (7)	0.4768 (7)	378	
C(2a)	-0.292 (2)	0.0439 (9)	0.3195 (7)	445	
C(2b)	0.117 (2)	-0.0058 (8)	0.2744 (7)	422	
C(3a)	-0.263 (2)	0.2696 (7)	0.1935 (7)	437	
C(3b)	0.165 (2)	0.2371 (9)	0.1754 (6)	465	
C(4)	0.035(1)	0.3042 (6)	0.3382 (5)	187	
C(5)	0.071 (1)	0.2764 (6)	0.4100 (5)	202	
C(6)	0.121 (1)	0.3402 (6)	0.4775 (5)	180	
C(7)	-0.010(1)	0.3689 (6)	0.5248 (6)	256	
C(8)	0.034 (2)	0.4291 (7)	0.5857 (6)	314	
C(9)	0.209 (2)	0.4605 (7)	0.5997 (6)	314	

C(10) C(11) C(12) C(13)	0.347 (1) 0.304 (1) 0.447 (1) 0.562 (2)	0.4324 0.3723 0.3440 0.3218	¥ (6) 3 (6) ) (7) 3 (8)	0.5532 (5) 0.4916 (5) 0.4435 (7) 0.4049 (7)	279 238 304 386			
Table 2. Geometric parameters (Å, °)								
$\begin{array}{c} Ge(1) - Ge(2)\\ Ge(1) - C(1)\\ Ge(1) - C(2)\\ Ge(2) - Ge(2)\\ Ge(2) - Ge(2)\\ Ge(2) - C(2)\\ Ge(3) - C(3)\\ Ge(3) - C(3)\\ Ge(3) - C(3)\\ Ge(3) - C(4)\\ C(1b) - Ge(2)\\ C(1b) - Ge(2)\\ Ge(2) - Ge(2)\\ Ge(2) - Ge(2)\\ Ge(2) - Ge(2)\\ Ge(1) - Ge(2)\\ Ge(1) - Ge(2)\\ Ge(1) - Ge(2)\\ Ge(3) $	Table 2. 2) a) b) b) c) c) c) c) c) c) c) c) c) c	Geometria 2.411 (3) 1.93 (1) 1.94 (1) 1.97 (9) 2.406 (2) 1.92 (1) 1.95 (1) 1.95 (1) 1.95 (1) 1.95 (1) 1.95 (1) 1.95 (2) 1.95 (3) 1.95 (4) 108.2 (5) 108.9 (4) 109.1 (5) 99.4 (3) 116.0 (4) 113.7 (4) 108.8 (4) 92.8 (1) 112.3 (5) 114.5 (4) 113.2 (4)	$\begin{array}{c} C(4) = 0 \\ C(5) = 0 \\ C(5) = 0 \\ C(6) = 0 \\ C(6) = 0 \\ C(7) = 0 \\ C(7) = 0 \\ C(7) = 0 \\ C(7) = 0 \\ C(10) = $	neters $(A, °)$ C(5) C(6) C(7) C(11) C(8) C(9) C(10) -C(11) -C(12) -C(13) -G(3)-C(3b) -C(4)-C(5) -C(5)-C(6) -C(5)-C(6) -C(5)-C(6) -C(5)-C(6) -C(5)-C(6) -C(5)-C(6) -C(5)-C(6) -C(5)-C(6) -C(5)-C(6) -C(5)-C(6) -C(5)-C(6) -C(7)-C(8) -C(7)-C(8) -C(9)-C(10) -C(10)-C(11)	$\begin{array}{c} 1.30 \ (1) \\ 1.52 \ (1) \\ 1.37 \ (1) \\ 1.37 \ (1) \\ 1.39 \ (1) \\ 1.35 \ (2) \\ 1.39 \ (2) \\ 1.39 \ (2) \\ 1.40 \ (1) \\ 1.44 \ (1) \\ 1.16 \ (2) \\ 109.0 \ (3) \\ 122.4 \ (7) \\ 122.0 \ (8) \\ 115.5 \ (6) \\ 120.1 \ (8) \\ 121.0 \ (8) \\ 118.9 \ (8) \\ 121 \ (1) \\ 120 \ (1) \\ 120 \ (1) \\ 120 \ (1) \end{array}$			
Ge(3)—Ge(3) Ge(2)—Ge(3) Ge(2)—Ge(3) Ge(2)—Ge(3) C(3a)—Ge(3)	$\begin{array}{c} C(2a) \\ C(2a) \\ C(2a) \\ C(3b) \\ C(3b) \\ C(3a) \\ C(3a) \\ C(3a) \\ C(4) \\ C(5) \\ C(5) \\ C(5) \\ C(6) \\ C(4) \\ C$	$\begin{array}{c} 113.2 \ (4) \\ 99.3 \ (3) \\ 114.6 \ (4) \\ 115.6 \ (4) \\ 108.9 \ (5) \\ 108.8 \ (4) \\ ) \\ -Ge(1) \\ -Ge(2) \\ -Ge(3) \\ -C(4) \\ -C(5) \\ -C(5) \\ -C(5) \\ -C(5) \\ -C(5) \\ -C(6) \\ -C(6) \\ -C(5) \\ -C(6) \\ -C(6) \\ -C(5) \\ -C(6) \\ -C(6)$	$\begin{array}{c} C(9) = (C_{1}) \\ C(6) = (C_{1}) \\ C(6) = (C_{1}) \\ C(1) = (C_{1}) \\ C(1) = (C_{1}) \\ C(1) = (C_{1}) \\ C(1) \\ C(1) \\ C(1) \\ C(1) \\ C(1) \\ C(1) \end{array}$	$\begin{array}{c} (10) - C(11) \\ C(11) - C(10) \\ C(11) - C(12) \\ C(11) - C(12) \\ C(12) - C(12) \\ C(12) - C(13) \\ \end{array}$	120 (1) 119.2 (9) 119.9 (9) 120.9 (9) 179 (1)			

The title compound was obtained from free singlet dimethylgermylene, thermally generated from 7-germanorbornadiene (Neumann, 1991), and 1,2-diethynylbenzene at 363 K in chlorobenzene (Weisbeck & Neumann, 1992). The air-sensitive crystals were obtained from methanol at 243 K, m.p. 365 K. The lattice parameters were determined from a symmetryconstrained least-squares fit. Refinement was based on fullmatrix least-squares methods with H atoms in calculated positions (C—H 0.96 Å).

Lists of structure factors, anisotropic thermal parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 55577 (18 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: AL1028]

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Spek, A. L. (1982). The EUCLID Package. In Computational Crystallography, edited by D. Sayre, p. 528. Oxford: Clarendon Press. Weisbeck, M. P. & Neumann, W. P. (1992). In preparation. quantum-mechanical determination of the strain enthalpy of a germirane shows a comparative instability of germiranes relative to the homologous cyclopropanes and siliranes (Horner, Grev & Schaefer, 1992). This may explain the experimental observation that no 1,2-addition product of germylenes to alkenes or dienes is formed.

0(10)

Me000

COOMe

cis isomer

66(1)

C(2o)

Ô

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# Structure of Dimethyl *trans*-3,3'-(1,1-Dimethylgermacyclopentane-3,4-diyl)bis[(*E*)-2-methylpropenoate]

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#### Abstract

This is the first crystal structure determination of a 3,4-dialkenyl-1-germacyclopentane. In the crystal this molecule resides around a twofold axis with Ge on this axis. The bond distances in the five-membered heterocycle are Ge—C 1.950(5), C—C 1.539(6), 1.550(6) Å and the exocyclic Ge—C bond distance is 1.929(4) Å. The ring torsion angles C—Ge—C—C, Ge—C—C—C, C—C—C—C—C are 14.1(3), 39.2(3),  $-53.2(4)^{\circ}$  and the bond angles around Ge are 92.1(2) (ring), 113.5(2), 110.6(2), 113.1(2)^{\circ}. Bond distances and angles in the alkenyl group are in the normal range.

#### Comment

This is the first example of an electron-deficient 1,3diene that reacts with dimethylgermylene to give a 3,4-dialkyenyl-1-germacyclopentane. A number of 1,3dienes, bearing no functional group (Neumann, Michels & Köcher, 1987; Bobbitt, Maloney & Gaspar, 1991) also give germacyclopentanes but no crystal structures have been obtained so far. The well known 1,2-cycloaddition of silvlenes to alkenes or dienes (Gaspar, 1978, 1981, 1985) has not yet been observed in the case of the heavier carbene analogue, the germylene. Styrene and its derivatives are the only alkenes that react with dimethylgermylene to give a cycloaddition product, which consists of two moles of the alkene and one mole of germylene. The reaction is assumed to proceed via an unstable 1:1 intermediate, a 2-phenylgermirane (Neumann, 1991). Conjugated 1,3dienes may give a 1,4-cycloaddition with Me<sub>2</sub>Ge or may react like styrene with only one double bond of the conjugated system to give 3,4-dialkenyl-1-germacyclopentanes with a molar ratio of diene/germylene of 2/1 (Neumann, 1991). The substitution pattern of methyl (E)-2-methyl-2,4-pentadienoate favours the latter reaction type, resulting in the formation of the title compound. Ab initio

Fig. 1. General view (*SHELXTL-Plus* graphic) of the molecule, showing the atom-numbering scheme. Anisotropic ellipsoids represent 50% probability boundaries. H atoms are represented as spheres of arbitrary radii.

## Experimental

CI8

Crystal data	
C <sub>16</sub> H <sub>26</sub> GeO <sub>4</sub> $M_r = 354.97$ Monoclinic C2/c a = 20.373 (9) Å b = 8.707 (4) Å c = 12.637 (6) Å $\beta = 126.91 (3)^\circ$ $V = 1792 (2) Å^3$ Z = 4	$D_x = 1.315 \text{ Mg m}^{-3}$ Mo K $\alpha$ radiation $\lambda = 0.71073 \text{ Å}$ Cell parameters from 16 reflections $\theta = 9.4-20.5^{\circ}$ $\mu = 1.70 \text{ mm}^{-1}$ T = 293 (1)  K Platelet 0.44 $\times$ -0.44 $\times$ 0.28 mm Colourless
Data collection Nicolet R3m/V diffractome- ter θ/2θ scans Absorption correction: none 2628 measured reflections 1171 independent reflections	$R_{int} = 0.0382$ $\theta_{max} = 22.5^{\circ}$ $h = -22 \rightarrow 22$ $k = 0 \rightarrow 10$ $l = -14 \rightarrow 14$ 6 standard reflections frequency: 150 min
997 observed reflections $[F > 4.0 \sigma(F)]$	intensity variation: none

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